# Synthesis of non-cross-linked polystyrene supported 5,5-dimethyl oxazolidinone chiral auxiliary 

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A new chiral auxiliary, non-cross-linked polystyrene supported 5,5-dimethyl oxazolidinone was synthesised in $25.1 \%$ yield using L-tyrosine as the starting material using a six-step method.

Keywords: non-cross-linked polystyrene, support, chiral auxiliary, 5,5-dimethyloxazolidinone

Oxazolidinone, the most versatile and widely used chiral auxiliary, was devised by Evans et al. ${ }^{1}$ This, so called, Evans' auxiliary has been utilised in a particularly wide variety of high diastereoselective reactions. ${ }^{2}$ However, the application of Evans' oxazolidinone is limited due to the drawbacks of the Evans methodology which involves the cleavage of the auxiliary. ${ }^{3}$ In order to address fully this troublesome cleavage problem, a new class of chiral auxiliary, the 5,5dimethyloxazolidinone has been developed. ${ }^{4}$ The improved auxiliary provides superior performance to many other chiral auxiliaries currently in common usage. It has proved to be particulary efficacious in terms of diastereoselectivity, yield and solubility. ${ }^{5-7}$

Recently, the attachment of Evans' oxazolidinone to polymersupport has been reported. ${ }^{8-9}$ The use of polymer-supported chiral auxiliaries gives some benefits: (1) easy separation and recovery of the expensive chiral material; (2) simple isolation of the desired chiral adduct; and (3) possible extension to a continuous system. But to the best of our knowledge, polymersupported 5,5 -dimethyloxazolidinone chiral auxiliary has not been reported. Our group has undertaken a research program to develop a new chiral auxiliary using non-cross-linked polystyrene (NCPS) as support. ${ }^{10}$ In our study another new chiral auxiliary based on NCPS was synthesised through six steps from the material of natural L-tyrosine (Scheme 1). It will be widely used in asymmetric synthesis, such as chiral alkylation, Aldol condensation, Diels-Alder and Michael addition.

## Experimental

Melting points were measured on a WRS-1A digital melting point apparatus and uncorrected; polarimetric analysis was recorded on a WZZ- $2 \mathrm{~B}^{\mathrm{A}}$ polarimeter; IR spectra were recorded on an IRspectrum One spectrometer (PE); ${ }^{1} \mathrm{H}$ NMR ( 600 MHz ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz ) spectra were recorded on a Varian Unity INOVA 600 spectrometer using TMS as internal standard; element analysis was determined by a VarioEL $\beta$ (Germany) analyser; MS spectra were recorded on Finnigan LCQ DUO MS system.

Synthesis of L-tyrosine methyl ester hydrochloride 1: Compound 1 was obtained as white crystals in yield (97\%). m.p. $189-190^{\circ} \mathrm{C}$ (lit. $\left.{ }^{[11]} 190-191^{\circ} \mathrm{C}\right),[\alpha]_{\mathrm{D}}{ }^{25}=+69.8\left(\mathrm{c} 1.06\right.$, pyridine) $\left[\right.$ lit. ${ }^{[11]}[\alpha]_{\mathrm{D}}{ }^{20}=+70.9$ (c 1.1, pyridine)].

Synthesis of N-Boc L-tyrosine methyl ester 2: Compound 2 can be synthesised according to ref.12. Yield $86 \%$. m.p. $104-105^{\circ} \mathrm{C}$ [lit. ${ }^{[13]}$ $\left.102^{\circ} \mathrm{C}\right],[\alpha]_{\mathrm{D}}{ }^{20}=+20.4$ (c 0.72 , THF) $\left[\right.$ lit. ${ }^{[13]}[\alpha]_{\mathrm{D}}{ }^{20}=-50.0$ (c 1.0, $\left.\left.\mathrm{CHCl}_{3}\right)\right]$. IR(NaCl): $v=3365,1740,1689 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}(600 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): 6.96(2 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}), 6.72(2 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H})$, $5.03(1 \mathrm{H}, \mathrm{m}, \mathrm{OH}), 4.54(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}), 3.71\left(3 \mathrm{H}, \mathrm{s}, \mathrm{O}-\mathrm{CH}_{3}\right), 3.01(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2}-\mathrm{Ar}\right), 2.98\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}-\mathrm{Ar}\right), 1.42\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}-\mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C}$ NMR ( 150 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $173.1,155.8,155.6,130.8(2 \mathrm{C}), 127.9,115.9(2 \mathrm{C}), 80.7$, $55.0,52.7,38.0,28.7(3 \mathrm{C})$; MS: $m / z=319.9\left(\mathrm{M}+\mathrm{Na}^{+}\right)$.

Synthesis of (3S)-3-(N-Boc-amino)-4-(4-hydroxyphenyl)-2-methyl-butan-2-ol 3: The compound $2(13.3 \mathrm{~g}, 45 \mathrm{mmol})$ in ether $(150 \mathrm{ml})$ was added dropwise to a solution of methylmagnesium iodide in ether $(300 \mathrm{ml})$ [From methyl iodide ( $28 \mathrm{ml}, 450 \mathrm{mmol}$ ) and magnesium ( $12 \mathrm{~g}, 500 \mathrm{mmol}$ )]. The mixture was stirred at room temperature for 48 h . Then the mixture was hydrolysed with ice and 2 M HCl until the magnesium was consumed. The ether layer was separated and the aqueous layer was extracted with ether ( $30 \mathrm{ml} \times 3$ ). The combined ether layer was dried over $\mathrm{MgSO}_{4}$ and evaporated to afford a pale


Scheme 1 Synthesis of NCPS supported 5,5-dimethyloxazolidinone.

[^0]yellow solid. Recrystallisation from $\mathrm{EtOH}-\mathrm{CH}_{3} \mathrm{COOEt}-\mathrm{H}_{2} \mathrm{O}(2: 3: 2$, $\mathrm{v} / \mathrm{v}$ ) gave 3 as colourless crystals ( $7.8 \mathrm{~g}, 58 \%$ ). m.p. $155.6-156.3^{\circ} \mathrm{C}$, $[\alpha]_{\mathrm{D}}{ }^{20}=-39.4$ (c 0.91, THF); IR(NaCl): $v=3355,1686 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 7.05(2H, d, $\left.J=7.8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}\right), 6.70(2 \mathrm{H}$, $\mathrm{d}, J=7.8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}), 5.67(1 \mathrm{H}, \mathrm{m}, \mathrm{OH}), 3.65(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}), 3.05(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2}-\mathrm{Ar}\right), 2.52\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}-\mathrm{Ar}\right), 1.20-1.18\left(15 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 155.7, 155.6, 130.8, 129.2(2C), 115.8(2C), 80.4, $72.4,68.2,37.4,28.8(3 \mathrm{C}), 26.7,21.2 ; \mathrm{MS}: m / z=318.1\left(\mathrm{M}+\mathrm{Na}^{+}\right)$. Elementary analysis calcd for compound 3: C, $65.06 \%, H, 8.53 \%$, N, $4.74 \%$; found: C, $65.18 \%$; H, $8.51 \%$; N, $4.59 \%$.

Synthesis of (S)-4-(4-hydroxybenzyl)-5,5-dimethyloxazolidinone 4. The $N$-Boc amino alcohol $3(7.5 \mathrm{~g}, 25.5 \mathrm{mmol})$ was suspended in $5 \% \mathrm{NaOH}(7.2 \mathrm{~g}, \mathrm{wt} \%)$ in $\mathrm{MeOH}(75 \mathrm{ml})$, refluxed for 12 h . The mixture was evaporated and the residue diluted with water $(20 \mathrm{ml})$ and ethyl acetate $(30 \mathrm{ml})$. The aqueous layer was extracted with ethyl acetate $(30 \mathrm{ml} \times 3)$. Then the combined organic phase was dried over $\mathrm{MgSO}_{4}$, concentrated under vacuum to give 4 ( $4.5 \mathrm{~g}, 81 \%$ ) as pale yellow solid which was used in the next step without further purification. m.p. $138.2-139.3^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}=-121.8$ (c 0.8, THF); IR(NaCl): $v=3290,1729,1515,1235 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 600 MHz , $\left.\mathrm{CH}_{3} \mathrm{OD}\right): 7.00(2 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}), 6.70(2 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}$, $\mathrm{Ar}-\mathrm{H}), 3.75(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}), 2.73\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}-\mathrm{Ar}\right), 2.62\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}-\right.$ Ar), $1.34\left(6 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(150 \mathrm{MHz}, \mathrm{CH}_{3} \mathrm{OD}\right): 160.0,156.3$, 130.2(2C), 128.3, 115.6(2C), 84.1, 63.5, 36.1, 26.9, 21.1; MS: m/z $=244.0\left(\mathrm{M}+\mathrm{Na}^{+}\right)$. Elementary analysis calcd for compound 4: C, $65.14 \%, \mathrm{H}, 6.83 \%, \mathrm{~N}, 6.33 \%$; found: C, $65.19 \%$; H, $6.82 \%$; N, 6.27\%.

Synthesis of (S)-4-[4-(4-vinylbenzyloxy)benzyl]-5,5-dimethyloxazolidinone 5: To a solution of $4(3.3 \mathrm{~g}, 15 \mathrm{mmol})$ in dry DMF ( 20 ml ) were added 4 -vinylbenzyl chloride (CMS) ( $2 \mathrm{ml}, 18 \mathrm{mmol}$ ), anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}(2 \mathrm{~g}, 15 \mathrm{mmol})$ and 18-crown-6 (catalyse amount). Then the resulting mixture was stirred for 12 h at r.t. The solvent was removed under vacuum and the residue diluted with $\mathrm{H}_{2} \mathrm{O}$ $(15 \mathrm{ml})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{ml})$. The aqueous layer was extracted with dichloromethane $(25 \mathrm{ml} \times 3)$ and the combined organic phase were dried over $\mathrm{MgSO}_{4}$ and evaporated to afford a pale yellow solid. Recrystallisation from $\mathrm{EtOH}-\mathrm{CH}_{3} \mathrm{COOEt}-\mathrm{H}_{2} \mathrm{O}(1: 2: 1$, v/v) give 5 (3.6 g, $72 \%$ ) as colourless crystals. m.p. $146.0-147.2^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}$ $=-53.65$ (c 0.58,THF); IR(NaCl): $v=3412,1781 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $7.44(2 \mathrm{H}, \mathrm{d}, J=7.8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}), 7.39(2 \mathrm{H}, \mathrm{d}$, $J=7.8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}), 7.09(2 \mathrm{H}, \mathrm{d},=7.2 \mathrm{~Hz}, \mathrm{Ar}), 6.93(2 \mathrm{H}, \mathrm{d}, J=7.2 \mathrm{~Hz}$, $\mathrm{Ar}), 6.75\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}_{1}=10.8 \mathrm{~Hz}, \mathrm{~J}_{2}=17.4 \mathrm{~Hz},=\mathrm{CH}\right), 5.78(1 \mathrm{H}, \mathrm{d}$, $\left.J=17.4 \mathrm{~Hz},=\mathrm{CH}_{2}\right), 5.27\left(1 \mathrm{H}, \mathrm{d}, J=10.8 \mathrm{~Hz},=\mathrm{CH}_{2}\right), 5.04(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{O}-\mathrm{CH}_{2}\right), 4.93(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 3.62(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}), 2.78\left(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{2}-\mathrm{Ar}\right)$, $2.63\left(1 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{2}-\mathrm{Ar}\right), 1.47-1.44\left(6 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C}$ NMR $(150 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): 158.4,158.2,137.8,136.8(2 \mathrm{C}), 130.3(2 \mathrm{C}), 129.5,128.1(2 \mathrm{C})$, $126.8(2 \mathrm{C}), 115.9(2 \mathrm{C}), 114.6,83.6,70.2,63.6,36.6,27.9,22.3$; MS: $m / z=337.9\left(\mathrm{M}^{+}\right)$. Elementary analysis calcd for compound 5:

C, $74.75 \%, \mathrm{H}, 6.87 \%, \mathrm{~N}, 4.15 \%$; found: C, $74.85 \%$; H, $6.89 \%$; N, 3.94\%.

Synthesis of NCPS supported 5,5-dimethyloxazolidinone 6: To a solution of monomer $5(3 \mathrm{~g}, 8.7 \mathrm{mmol})$ in THF ( 15 ml ) were added styrene $(1.85 \mathrm{~g}, 17.4 \mathrm{mmol})$ and $\operatorname{AIBN}(0.01 \mathrm{~g}, 0.06 \mathrm{mmol})$. The mixture was stirred at $70^{\circ} \mathrm{C}$ for 48 h under nitrogen atmosphere. Then most of the solvent was removed under reduced pressure. The viscous solution was dropped into cold ethanol ( 200 ml ) and the precipitated solid was filtered and dried at $65^{\circ} \mathrm{C}$ for 2 h under vacuum to afford solid polymer 6 ( $4.32 \mathrm{~g}, 89 \%$ ). IR(NaCl): $\mathrm{v}=3280,1752 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $7.25-6.45(\mathrm{~m}$, broad signal due to the polymer resonance), $4.90\left(2 \mathrm{H}, \mathrm{s}, \mathrm{O}-\mathrm{CH}_{2}\right), 3.62(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}), 2.75$ $\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}-\mathrm{Ar}\right), 2.60\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}-\mathrm{Ar}\right), 1.81-1.25(\mathrm{~m}$ broad signal due to the polymer resonance); ${ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : 158.4 , $145.5,130.4,129.5,128.1,126.2,115.8,83.6,70.5,63.5,40.8,36.6$, 28.0, 22.4. Elementary analysis calcd for polymer 6: C, $81.43 \%$; H, $7.20 \%$; N, $2.57 \%$; found: C, $81.83 \%$; H, $7.22 \%$; N, $2.48 \%$.

NCPS supported 5,5-dimethyl oxazolidinone chiral auxiliary was characterised by IR, ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, GPC, TGA and DSC. The result showed that the obtained polymer has high loading capacity, remarkable solubility, low molecule and good thermally stability. The overall yield reached $25.1 \%$.

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